

Trimethylsilyl rifluoromethanesulfonate

Versatile Reagent and Catalyst

Trimethylsilyl trifluoromethanesulfonate(1, trimethylsilyl triflate, TMSOTf) is an exceptionally strong silvlating

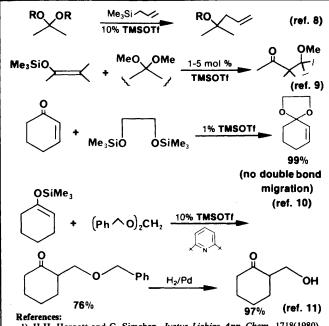
agent.1 Representative reactions in which 1 is used stoichiometrically are shown below.

$$O = \frac{2 \text{ TMSOT!}}{\text{Et}_3 \text{N}} OSiMe_3$$
 (ref. 2)

$$t$$
-Bu t -Bu

Trimethylsilyl trifluoromethanesulfonate also functions as a catalyst for a variety of transformations of importance in organic synthesis. The versatility of 1 in this regard is illustrated below.

MeO OMe
$$\frac{Me_3SiH}{1\% \text{ TMSOTf}}$$
 MeO H (ref. 7)



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Trimethylsilyl trifluoromethanesulfonate

10g \$8.00; 50g \$30.00

Triethylamine......500g \$8.75; 2kg \$22.45 13,206-3

1,8-Diazabicyclo[5.4.0]undec-7-ene (DBU) 13,900-9

25g \$9.45; 100g \$26.30; 500g \$87.75

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22,580-0

50g \$22.50

21,958-4 2,6-Di-tert-butylpyridine5g \$19.60



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TRIMETHYLBROMOSILANE TRIMETHYLIODOSILANE TRIMETHYLSILYL TRIFLUOROMETHANESULFONATE

(CH₃)₃SiBr

CF₃SO₃Si(CH₃)₃

(CH₃)₃Sil

Three Highly Reactive Silylating Agents

The silylating rates of ten different silylating agents were investigated by H.H. Hergott and G. Simchen¹⁾: trimethylbromosilane, trimethylsilyl trifluoromethanesulfonate and trimethyliodosilane gave by far the highest reaction rates.

Relative rates (k_{rel}) for the silylation of ketones by some silylating agents [(CH₃)₃SiX]:

Х	CI	CH₃SO₃	C ₆ H₅SO₃	(CH ₃) ₃ SiOSO ₃	Br	CF₃SO₃	ı
k _{rel}	1	40	160	270	7.9.10⁴	6.7.10 ⁸	7.10°

Trimethylbromosilane²⁾, trimethylsilyl trifluoromethanesulfonate^{3) 4) and ref in 1)} and trimethyliodosilane⁵⁾ were also described by other authors as reagents for silylating reactions. For further synthetic uses of these highly reactive reagents note the following references and the references cited therein [a comprehensive review on $(CH_3)_3SiBr$ and $(CH_3)_3SiB$

(CH₃)₃SiBr⁶⁻⁸⁾²¹⁾²²⁾

e.g cleavage of ethers 6), lactones 8), orthoesters 6) and phosphonates 7); conversion of alcohols to bromides 6); synthesis of glycosyl bromides 21)

CF₃SO₃Si(CH₃)₃¹⁾⁹⁻¹⁴⁾

e.g. cleavage of oxiranes⁹⁾¹⁰⁾; catalyst for different reactions: synthesis of nucleosides, conversion of acetals to ethers¹¹⁾, acetalization under aprotic conditions¹¹⁾, condensation of silyl enol ethers with acetals¹²⁾¹³⁾; cleavage of tert-butylesters under non acidic conditions¹⁴⁾

 $(CH_3)_3Sil^{8)15-22}$

e.g. cleavage of ethers¹⁵⁾, esters, lactones⁸⁾, acetals¹⁶⁾, phosphonates¹⁷⁾ and carbamates; deoxygenation of sulfoxides and α-ketols¹⁸⁾; conversion of alcohols to iodides; conversion of ketals to ketones; synthesis of allylic alcohols¹⁹⁾; synthesis of α-iodosulfides²⁰⁾

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initial salary in the range £6070 -£6880 per annum.

Applications should be received as soon as possible by The Registrar, The University, P.O. Box 147, Liverpool

Quote Ref: RV/792/CC.

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Department of Organic Chemistry

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